## UNIVERSITY OF MYSORE <br> Postgraduate Entrance Examination November - 2021



(Read carefully the instructions given in the Question Booklet)

## course : M.Tech. <br> SUBJECT : <br> MATERIALS SCIENCE

MAXIMUM MARKS : 50
(Including time for filling O.M.R. Answer sheet)

## INSTRUCTIONS TO THE CANDIDATES

1. The sealed question paper booklet containing 50 questions enclosed with O.M.R. Answer Sheet is given to you.
2. Verify whether the given question booklet is of the same subject which you have opted for examination.
3. Open the question paper seal carefully and take out the enclosed O.M.R. Answer Sheet outside the question booklet and fill up the general information in the O.M.R. Answer sheet. If you fail to fill up the details in the form as instructed, you will be personally responsible for consequences arising during evaluating your Answer Sheet.
4. During the examination:
a) Read each question carefully.
b) Determine the Most appropriate/correct answer from the four available choices given under each question.
c) Completely darken the relevant circle against the Question in the O.M.R. Answer Sheet. For example, in the question paper if "C" is correct answer for Question No.8, then darken against SI. No. 8 of O.M.R. Answer Sheet using Blue/Black Ball Point Pen as follows:

> Question No. 8. (A) (B) (D) (Only example) (Use Ball Pen only)
5. Rough work should be done only on the blank space provided in the Question Booklet. Rough work should not be done on the O.M.R. Answer Sheet.
6. If more than one circle is darkened for a given question, such answer is treated as wrong and no mark will be given. See the example in the O.M.R. Sheet.
7. The candidate and the Room Supervisor should sign in the O.M.R. Sheet at the specified place.
8. Candidate should return the original O.M.R. Answer Sheet and the university copy to the Room Supervisor after the examination.
9. Candidate can carry the question booklet and the candidate copy of the O.M.R. Sheet.
10. The calculator, pager and mobile phone are not allowed inside the examination hall.
11. If a candidate is found committing malpractice, such a candidate shall not be considered for admission to the course and action against such candidate will be taken as per rules.
12. Candidates have to get qualified in the respective entrance examination by securing a minimum of 8 marks in case of SC/ST/Cat-I Candidates, 9 marks in case of OBC Candidates and 10 marks in case of other Candidates out of 50 marks.

## INSTRUCTIONS TO FILL UP THE O.M.R. SHEET

1. There is only one most appropriate/correct answer for each question.
2. For each question, only one circle must be darkened with BLUE or BLACK ball point pen only. Do not try to alter it.
3. Circle should be darkened completely so that the alphabet inside it is not visible.
4. Do not make any unnecessary marks on O.M.R. Sheet.
5. Mention the number of questions answered in the appropriate space provided in the O.M.R. sheet otherwise O.M.R. sheet will not be subjected for evaluation.

1) X-ray crystallography uses which characteristic of light?
(A) polarization
(B) interference
(C) diffraction
(D) coherency
2) Which of the following contribute to the reason behind the origin of surface tension?
(A) only cohesive forces
(B) only adhesive forces
(C) neither cohesive forces nor adhesive forces
(D) both cohesive forces and adhesive forces
3) Which of the following is true?
(A) for an isolated system, $\mathrm{dS}>=0$
(B) for a reversible process, $\mathrm{dS}=0$
(C) for an irreversible process, $\mathrm{dS}>0$
(D) all of the mentioned
4) Which of the following is the wave number of near infrared spectrometer?
(A) $4000-200 \mathrm{~cm}^{-1}$
(B) $200-10 \mathrm{~cm}^{-1}$
(C) $12500-4000 \mathrm{~cm}^{-1}$
(D) $50-1000 \mathrm{~cm}^{-17}$
5) In Raman spectroscopy, the radiation lies in the $\qquad$ .
(A) microwave region
(B) visible region
(C) uv region
(D) x-ray region
6) The crystal lattice has a $\qquad$ arrangement.
(A) one-dimensional
(B) two-dimensional
(C) three-dimensional
(D) four-dimensional
7) What happens to the electropositive character of elements on moving from left to right in a periodic table?
(A) increase
(B) decreases
(C) first increases than decreases
(D) first decreases than increases
8) How many 'd' electrons are present in $\mathrm{Cr}^{2+}$ ion
(A) 4
(B) 5
(C) 6
(D) 3
9) Give the IUPAC name for the compound :
(A) 2-choro-3-methylbutane
(B) 2-choro-3-methylbutene
(C) 3-choro-3-methylbutane
(D) 3-choro-2-methylbutane
10) When phenol is treated with excess of bromine water, it gives which of the following product?
(A) m-bromophenol
(B) o-and p-bromophenol
(C) 2,4-dibromophenol
(D) 2,4,6-tribromophenol
11) The application of color to the whole body of a textile material with somedegree of fastness
(A) dyeing
(B) printing
(C) discharge style
(D) none
12) If the temperature of any gas is increased, its volume
(A) increases
(B) decreases
(C) remains same
(D) none of above
13) Ziegler-Natta catalysis is associated with:
(A) alkene hydrogenation
(B) alkene polymerization
(C) hydroformylation of alkenes
(D) alkyne metathesis
14) Which of the following polymer type is not classified on the basis of its application and Properties?
(A) rubbers
(B) plastics
(C) fibres
(D) synthetic
15) Which of the following will decrease the rate of reaction?
(A) catalytic poison
(B) positive catalyst
(C) negative catalyst
(D) catalytic promoters
16) Corrosion of metals involves
(A) physical reactions
(B) chemical reactions
(C) both
(D) none
17) If $A$ and $B$ matrices are of same order and $A+B=B+A$, this law is known as
(A) distributive law
(B) commutative law
(C) associative law
(D) cramer's law
18) Circumference of the circle is calculated by
(A) $2 \pi r$
(B) $2 \pi / \mathrm{r}$
(C) $\pi r / 2$
(D) $\pi r$
19) Circumcentre of the triangle, whose vertices are $(0,0),(6,0)$ and $(0,4)$ is
(A) $(3,2)$
(B) $(2,0)$
(C) $(0,3)$
(D) $(3,0)$
20) The value of $\sin -1(\sin 12)+\cos -1(\cos 12)$ is equal to
(A) zero
(B) $24-2 \mathrm{p}$
(C) $4 p-24$
(D) none of these
21) Which of the following viruses have a complex symmetry?
(A) alphavirus
(B) mobillivirus
(C) orthopoxvirus
(D) parvovirus
22) Which of the following is also known as invert sugar?
(A) sucrose
(B) fructose
(C) dextrose
(D) glucose
23) Which of these in not a lipid
(A) fats
(B) oils
(C) proteins
(D) waxes
24) The word $\qquad$ comes from the name of a Persian mathematician Abu Ja'far Mohammed ibn-i Musa al Khowarizmi.
(A) flowchart
(B) flow
(C) algorithm
(D) syntax
25) In double precision format, the size of the mantissa is $\qquad$ .
(A) 32 bit
(B) 52 bit
(C) 64 bit
(D) 72 bit
26) SYS command is used to
(A) copy dos system files to new disk
(B) copy dos configuration files to a new disk
(C) update the dos system files
(D) none of above
27) Which statement is incorrect?
(A) at constant pressure, $\mathrm{h}=\mathrm{e}+\mathrm{Pv}$
(B) the thermodynamic symbol for entropy is s
(C) gibbs free energy is a state function
(D) for an endothermic process, $h$ is negative
28) Change in enthalpy of a system is due to heat supplied at
(A) constant volume
(B) constant pressure
(C) both at constant volume and pressure
(D) none of the mentioned
29) The entropy of an isolated system can never $\qquad$ .
(A) increase
(B) decrease
(C) be zero
(D) none of the mentioned
30) What is the correct increasing order of stretching frequencies for $\mathrm{C} \equiv \mathrm{C}$, $\mathrm{C}=\mathrm{C}$ and $\mathrm{C}-\mathrm{C}$ ?
(A) $\mathrm{C}-\mathrm{C}>\mathrm{C}=\mathrm{C}>\mathrm{C} \equiv \mathrm{C}$
(B) $\mathrm{C} \equiv \mathrm{C}>\mathrm{C}=\mathrm{C}>\mathrm{C}-\mathrm{C}$
(C) $\mathrm{C}-\mathrm{C}>\mathrm{C}=\mathrm{C}<\mathrm{C} \equiv \mathrm{C}$
(D) $\mathrm{C} \equiv \mathrm{C}<\mathrm{C}-\mathrm{C}>\mathrm{C}=\mathrm{C}$
31) Sky looks blue because the sun light is subjected to $\qquad$ .
(A) rayleigh scattering
(B) compton scattering
(C) both
(D) none
32) Which of the following defines the Mass number of an atom?
(A) number of protons + number of electrons
(B) number of neutrons + number of electrons
(C) number of protons + number of neutrons
(D) number of electrons
33) Which of the following is the correct order of the atomic radii of the elements oxygen, fluorine and nitrogen?
(A) O $<$ F $<$ N
(B) $\mathrm{N}<$ F $<$ O
(C) O $<$ N $<$ F
(D) F $<$ O $<$ N
34) What happens to the electropositive character of elements on moving from left to right in a periodic table?
(A) increase
(B) decrease
(C) first increases than decreases
(D) first decreases than increases
35) Choose the correct one which will react faster in the SN2 nucleophilic substitution reaction
(A) $\mathrm{CH}_{2}-\mathrm{CH}=\mathrm{CH}_{2}=\mathrm{Br}$
(B) $\mathrm{CH}_{2}-\mathrm{CH}=\mathrm{CH}_{2}-\mathrm{Br}$
(C) $\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2}=\mathrm{Br}$
(D) $\mathrm{CH}=\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{Br}$
36) Which of the following regents may be used to distinguish between phenol and benzoic acid?
(A) aqueous NaOH
(B) tollen's reagent
(C) molisch reagent
(D) neutral $\mathrm{FeCl}_{3}$
37) In Avagadro's law, volume is directly proportional to :
(A) pressure
(B) temperature
(C) number of moles
(D) volume
38) Corrosion of metals involves
(A) physical reactions
(B) chemical reactions
(C) both
(D) none
39) The concept of matter wave was suggested by
(A) heisenberg
(B) de broglie
(C) schrodinger
(D) laplace
40) What is the size of colloidal particles?
(A) $10-20 \mathrm{~nm}$
(B) more than 20 nm
(C) less than 10 nm
(D) 30 to 50 nm
41) We can add two matrices having real numbers $A$ and $B$ if their
(A) order is same
(B) rows are same
(C) columns are same
(D) elements are same
42) The differential equation $2 \frac{d y}{d x}+x^{2} y 2 x+3, y=(0)=5$ is
(A) linear
(B) nonlinear
(C) linear with fixed constants
(D) undeterminable to be linear or nonlinear
43) If the distance from the focus is 3 units and the distance from the directrix is 3 units, then how much is the eccentricity?
(A) infinity
(B) zero
(C) unity
(D) less than one
44) Peptide bond is a $\qquad$ .
(A) covalent bond
(B) ionic bond
(C) metallic bond
(D) hydrogen bond
45) The degree of unsaturation of lipids can be measured as
(A) iodine number
(B) saponification number
(C) reichert meissel number
(D) polenske number
46) Vitamin $C$ is present in :
(A) tomatoes
(B) papaya
(C) guava
(D) all of the above
47) Name the amino acid which is optically inactive
(A) glycine
(B) alanine
(C) valine
(D) proline
48) What is the first step in photosynthesis?
(A) generation of ATP
(B) formation of NADPH
(C) Through light, excitement of an electron of chlorophyll pigment.
(D) formation of glucose
49) If $\mathrm{a}, \mathrm{b}$ and c are integers, then according to associative law of multiplication the $(a \times b) \times$ cmust be equal to
(A) $\mathrm{a} \times(\mathrm{b}+\mathrm{c})$
(B) $(a-b) \times c$
(C) $(\mathrm{a}+\mathrm{b})+\mathrm{c}$
(D) $\mathrm{a} \times \mathrm{b}+\mathrm{a} \times \mathrm{c}$
50) The product of a rational and an irrational numbers is :
(A) always an integer
(B) always a rational number
(C) always an irrational number
(D) sometimes rational and sometimes irrational

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## Rough Work

## అభ్యథిรగษిగి శ్జอఒసేగఆు



 ఎంబదన్ను யరిరిలలిసిరి.



 జదాబ్దారరంగిరుత్తిర.


 లుత్తరహస్ను నిధణరిి.


 కుంబిర:




 ळలళెయల్లిన లుదాळరణ నైలణి.
 యృడ్బొలు.
 పిల్టలిద్యానిలయుద
 ஹృఁగబळుదు.




 అంచగఆన్ను யֹడియత్ర్ప్దు.

## ఓ.ఎం.ఆరా. ळలఆయన్ను కుంబలు ష్యృజసెగళు









Note : English version of the instructions is printed on the front cover of this booklet.

